Accelerating the Computation of Detailed Chemical Reaction Kinetics for Simulating Combustion of Complex Fuels



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Motivation: Changing World of Fuels and Engines

- Fuel streams are rapidly evolving
 - Heavy hydrocarbons
 - Oil sands
 - Oil shale
 - Coal
 - New renewable fuel sources
 - Ethanol
 - Biodiesel
- New engine technologies
 - Direct Injection (DI

- Homogeneous Charge Compression Ignition (HCCI)
- Low-temperature combustion
- New mixed modes of combustion (dilute, high-pressure, low-temp.)
- Sound scientific understanding is necessary to develop predictive, validated multi-scale models!





Combustion chemistry

• Example, natural gas combustion

 $CH_4 + 2O_2 => CO_2 + H_2O_2$

- Occurs through a reaction network producing and consuming intermediate species
 - CO, OH, H_2O_2 , HO_2 , CH_3 , ...
- Detailed chemical mechanisms are needed to compute
 - Flame structure and stability
 - Emissions

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Validate reduced reaction mechanisms



Detailed chemical kinetics are expensive

- Chemical source term evaluation is computationally intensive
- Thousands of elementary reaction steps accumulated to global species reaction rates
- Often the target for model reductions or algorithmic improvements
- How fast can we compute detailed chemical kinetics on accelerators?

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Chemistry Kernels

- Reaction rates, thermodynamic properties and transport coefficients account for 55% of time.
 - Complex chemical kinetic models needed to address multi-stage ignition and flame dynamics
- Point-wise functions that are independent of DNS software's mesh data structure and MPI-layer
 - Uses Chemkin API
- Used across other combustion codes in the community.
 - Impacts other HPC and workstation-scale combustion applications.
- Accelerator library targets the DNS chemistry needs and beyond

Kyle Spafford (ORNL) et al., "Accelerating S3D: A GPGPU Case Study," in Seventh International Workshop on Algorithms, Models, and Tools for Parallel Computing on Heterogeneous Platforms (HeteroPar 2009). Delft, The Netherlands, 2009







Accelerator library for combustion kinetics

• Conservation equation in a typical combustion application

$$\frac{\partial Q}{\partial t} + u \frac{\partial Q}{\partial x} + \dots = \dot{\omega}_Q$$

$$\dot{\omega}_Q = f(\dot{\omega}_C)$$

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where $\dot{\omega}_{c}$ is the rate of chemical kinetics

- Chemistry kernel evaluates the chemical kinetics for large mechanisms.
- Well optimized on CPUs and achieves more than 20% of peak on AMD opterons
- Porting to GPU and larger chemistry requires higher levels of parallelism



Parallelizing reaction kinetics (CKWYP)

$$\dot{\omega}_{M} = \sum_{N} \Omega(P, T, Y_{M})$$

- Grid-level parallelism (several independent states)
 - Will provide MPI parallelism
 - In some cases, also SMP-like parallelism
- Grid-level vectorization does not provide sufficient performance
 32(states) * 4000 variables * 8bytes = 1000 kB
- Current capacity in shared memory/L1 cache = 64kB
- Need to go deeper for vector parallelism
 - Equation level parallelism



Data flow in the rates kernel



- Data movement should be minimized while also vectorizing
- Expose concurrency (independent blocks) within the reaction network
- Redundant computation to achieve parallelism
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Partitioning at species/reaction level

- Similar to partitioning the grid for distributed memory parallelism (MPI)
- Why partition the computation at species/reaction level?
 - Asynchronous execution to hide latencies and data transfers (memcpy across PCI)
 - Distribute work to multiple accelerators assigned to a single host
 - Allow finer grained parallelism at the chemistry level to multiply the scalability of the flow solver
- Keiki treats the chemical kinetics as a graph and partitions it to minimize edgecut and maximize parallel performance



Reaction network as a graph

- Chemical reaction network is a bi-partite graph between two sets of vertices
 - The species form one set
 - The reactions form the second set
 - Stoichiometry of the reaction network defines the graph
- The adjacency matrix of the graph is

$$A = \left(\begin{array}{cc} 0 & B \\ B^T & 0 \end{array}\right)$$

Where B is the M x N stoichiometry matrix

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Partitioning the graph

- Graph partitioning software Metis and PaToH were used to partition the bi-partite graph
 - A good quality partition minimizes edge-cut with maximum load balance
 - Reorders the network, without changing the answers
- Edge-cut induces redundant computation or synchronization points
- Partitions should be sized to meet the vector length and memory requirement
 - Large enough to have enough number of threads per thread block
 - Control shared memory requirement to obtain high occupancy
- Need a sufficient number of partitions that can execute concurrently



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Partitioning iso-octane chemistry

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• LLNL's detailed mechanism for gasoline surrogate composed of 858 species and 3606 reactions





Partitioning iso-octane chemistry





• The quality of partitioning gets better as the chemistry model gets bigger



Keiki – Code Generator

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Performance results

- Performance on dual 6-core Opteron CPU and Fermi GPU were compared for 52-species n-heptane and 858 species iso-octane chemistry
 - CPU peak = 2*62.4 = 125 GF
 - GPU peak = 515 GF
- The CPU code was well optimized and tuned for performance
- The execution times on GPU were 3X faster than the CPU
- Work in progress to measure and tune performance on Kepler





GPU library coupled to combustion CFD

• Work in progress:

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- A flamelet equation solver is being developed around the CUDA library
- CUDA library for chemical kinetics is being coupled to Forte in partnership with Reaction Design
 - Forte ported to Jaguar (Cray XK6)
 - Software linking and API are being explored









Summary

- New software and techniques were developed to enable the computation of combustion chemistry on GPU accelerators using the CUDA programming model
- Significant potential to accelerate the computation of very large detailed mechanisms
- What started out as an effort to accelerate S3D has been extended to much larger chemical mechanisms.

